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SARDAR PATEL UNIVERSITY

M. Sc. Physics Ist Semester Examination

Monday, Date: 11-04-2016 Time: 10.30 AM to 01.30 PM

CBCS Course No.: PS01CPHY03
Subject: Atomic Molecular & Laser Physics

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Note: Symbols have their usual meaning. Total Marks: 76		
Q.1		rite answers of all eight questions in a table form by showing your choice (8) ainst the question number.
	(1)	In the case of H-atom "centrifugal barrier potential" varies as
		(a) $\frac{1}{r}$ (b) $\frac{1}{r^2}$ (c) $\frac{1}{r^3}$ (d) r^2
	(2)	For three level laser system, gives necessary mathematical condition
	` ,	for population inversion between level 1 and 2.
		(a) $T_{32} = T_{21}$ (b) $T_{32} > T_{21}$ (c) $T_{32} < T_{21}$ (d) $T_{32} > T_{31}$
	(3)	Most probable decay mechanism of 2S _{1/2} is
		(a) non-radiative transition (b) one-photon spontaneous radiation
		(c) two-photon emission (d) completely forbidden
	(4)	Life time for cavity photon in a given lasing system depends on
		(a) internal absorption (b) leakage through window (mirror-2)
		(c) scattering (d) all of the above
	(5)	gas is used to increase the efficiency of the CO ₂ gas laser.
		(a) He (b) N_2 (c) O_2 (d) NH_3
	(6)	The energy difference between various vibrational levels corresponds to the
		region
		(a) infrared (b) visible (c) ultraviolet (d) far infrared
	(7)	Angle between HOH in H_2O molecule is more than 90^0 , because of
		(a) repulsion between electrons (b) stretching effect of rotation
		(c) hybridization effect of 2s and 2p orbitals (d) dipolar nature of H ₂ O
	(8)	- 'W
		(a) ESR (b) hyperfine structure (c) Lamb shift (d) Lyman-α line
Q.2	Λι	nswer any seven questions. (14)
	(1)	"Rydberg atoms can be treated as hydrogenic atoms" - why?
	(2)	Obtain the eigenvalue of permutation or interchange operator.
	(3)	With usual notation, prove that for two electron system, $S_{i}\chi_{i}(1,2) = \chi_{1}(1,2)$.
	(4) (5)	Write two assumptions involved in Thomas-Fermi model for many-electron atom. Draw a simple diagram showing a few rotational levels for the ground vibrational
	(J)	states in a molecule.
	(6)	Why normal optical sources do not emit stimulated light?

Give difference between Rayleigh and Raman scattering. (9)Briefly describe the basic principle of free electron laser. For H-atom, setup Hamiltonian and derive $\left[\frac{d^2}{d\rho^2} - \frac{l(l+1)}{\rho^2} + \frac{\lambda}{\rho} - \frac{1}{4}\right] u_{E,l}(\rho) = 0$. Q.3 (a) (6)Here, $\rho = \left(-\frac{8\mu E}{\hbar^2}\right)^{-\frac{1}{2}}r$ and $\lambda = \frac{Ze^2}{4\pi\epsilon_0\hbar}\left(-\frac{\mu}{2E}\right)^{-\frac{1}{2}}$ for bound state (E < 0). Write detailed note on Lamb shift experiment. **(b)** (6)Write expression for Hamiltonian for two-electron system. Based on the Pauli's (6) exclusion principle, discuss the symmetric and antisymmetric properties of wave function. Q.4 (a) Derive $\frac{d^2\chi(x)}{d\chi^2(x)} - \frac{1}{\sqrt{x}} [\chi(x)]^{\frac{3}{2}} = 0$ using Thomas-Fermi theory for many-electron **(6)** atoms. Write at least one limitation of the theory. Give detailed note on LCAO method considering an example of H₂⁺ ion. (6) OR **(b)** Based on Born-Oppenheimer approximation, derive an equation for total **(6)** energy (Es,v,j) for diatomic molecule. Discuss the use of Morse potential in determining E_{s,v,f}. Q.5 (a) Derive necessary equations to show how variation of Laser power around (6) threshold condition take place. **(b)** For three-level laser system, obtain the condition for population inversion, and (6)derive an expression for threshold pump power. Average pump frequency of a given three-level Ruby laser is 6.25×10¹⁴Hz. Density of Cr⁺³ ions and threshold pump rate (W_{pt}) are 1.6×10^{19} cc and 330 s⁻¹. Calculate threshold pump power. Using an expression $\Gamma_{12} = \frac{1}{2\epsilon_0} \frac{D_{21}^2}{h^2} \int u(\omega) \left\{ \frac{\sin[(\omega_{21} - \omega)/2]t}{(\omega_{21} - \omega)/2} \right\}^2 d\omega$ for transition **(6)** probability and assuming $u(\omega)$ varies slowly, obtain expressions for Einstein coefficients B_{12} and A. Q.6 (a) Write detailed note on semiconductor laser. **(6)** Write the basic principle of laser. Discuss the working of He-Ne laser with (b) (6)the help of suitable diagram. Why only specific dimension of He-Ne discharge tube is selected? OR

(7)

(8)

What are cold atoms?

Write detailed note on NII3 maser.

(6)